



10th Brazilian Conference on Natural Products

XXXVI RESEM

4-7 November 2025, Belo Horizonte, MG, Brazil



Section: 02

COMPARATIVE EXTRACTION OF COSTIC ACID FROM *Nectandra barbellata* - CONVENTIONAL vs. SUPERCRITICAL CO₂ METHODS

Emmanuele Di Fabio^{1*}, Dalete C. S. Souza¹, Marina G. Monroe¹, Roberto B. P. Almeida², Priscila C. Vegg³, Juliana M. Prado⁴, João Henrique G. Lago¹, Bruno G. da Silva¹.

e.fabio@ufabc.edu.br

1-Center for Natural and Human Sciences, Federal University of the ABC, Santo André 09210-170, SP, Brazil.

2- Institute of Biosciences, University of São Paulo, São Paulo 05508-090, SP, Brazil. 3- Department of Chemical Engineering, Institute of Environmental, Chemical and Pharmaceutical Sciences, Federal University of São Paulo, Diadema 09913-030, SP, Brazil. 4- Engineering Modeling and Applied Social Sciences Center, Federal University of ABC, Santo André 09210-580, SP, Brazil.

In alignment with the guidelines that regulate the limits of residual organic solvents in pharmaceutical products (ICH Q3C)¹, this study evaluated the environmentally friendly approach of supercritical CO₂ for the extraction of costic acid, a promising antiparasitic - against the protozoal *Trypanosoma cruzi* - sesquiterpene from *Nectandra barbellata* (Lauraceae)², comparing it to the conventional maceration technique. In the maceration process, 16 g of dried plant material were extracted with 50 mL of solvent (hexane or EtOH) in six successive steps. For supercritical extraction, the conditions of 47.5 °C and 35 MPa were selected based on prior experimental design. After solvents removal under vacuum, the total yield and the costic acid content, quantified by HPLC-DAD, were determined. The results are summarized in Table 1. Supercritical fluid extraction (SFE) exhibited better selectivity, achieving the highest concentration of costic acid in the extracts (450 mg/g). Its productivity was markedly superior to conventional extraction methods, reflecting enhanced process efficiency. This superior performance, combined with reduced use of organic solvents, aligns SFE with green chemistry principles and supports more efficient, sustainable, and environmentally friendly processes to obtain this bioactive compound.

Table 1: Comparison among the yield and costic acid content of *N. barbellata* aerials parts extracts obtained by SFE and maceration techniques.

Extraction method	Yield (%), d.b.)	CAC (mg/g extract)	CAC (mg/g RM d.b.)	Time (h)	Productivity (g h ⁻¹)
SFE	4.0 ± 0.2	450	18.01	0.8	0.4655
HEX	5.2 ± 0.2	358	18.71	240	0.0034
EtOH	12.2 ± 0.4	318	38.84	240	0.0078

RM raw material, CAC costic acid content, SFE extract obtained using supercritical fluid extraction, HEX extract obtained using hexane (maceration), EtOH extract obtained using ethanol (maceration), d.b. dry basis.

Keywords: Supercritical fluid extraction, costic acid quantification, green chemistry, sustainable extraction methods.

References: ¹ICH Q3C(R9), 2024. ²Lonero, V. S. *et al.*, *Bioorganic Chemistry*, **2019**, 95, 103510.

Acknowledgments: CAPES, CNPq, FAPESP.



Sociedade Brasileira de Química
Divisão de Produtos Naturais

